organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.010 Å Disorder in main residue R factor = 0.085 wR factor = 0.244 Data-to-parameter ratio = 9.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

A synthetic intermediate of enantiopure *N*-protected β -hydroxyvaline, (2*R*)-2-[*N*-(*tert*-butoxycarbonyl)-amino]-3-methylbutane-1,3-diol

Crystals of the title compound [systematic name: (2R)-tertbutyl N-(1,3-dihydroxy-3-methyl-2-butyl)carbamate], C₁₀H₂₁-NO₄, were successfully grown from an ethyl acetate solution. There is one independent molecule in the asymmetric unit. The urethane linkage, -O-CO-NH-, is a *cis* conformer and the molecules are linked into a dimer by two N-H···O=C hydrogen bonds. The dimers are connected together by O-H···O hydrogen bonds in which four hydroxy groups are involved.

Comment

In need of enantiopure β -hydroxyvaline-containing peptide antibiotics, a chiral form of 2-*N*-(*tert*-butoxycarbonyl)amino-3methylbutane-1,3-diol was recently reported as an effective intermediate for scale-up synthesis (Dettwiler & Lubell, 2003). Among various peptide antibiotics, β -hydroxyvaline is found, as in aureobasidines (Takesako *et al.*, 1991) and luzopeptines (Konishi *et al.*, 1981). In this paper, we report the structure of the title compound, (I), as one of our synthetic studies of antibacterial peptides containing unusual amino acids (Oku *et al.*, 2004; Yamada *et al.*, 2004; Urakawa *et al.*, 2004) and *N*-methylated amino acids (Endo *et al.*, 2003).

$$\begin{array}{ccc} \mathsf{CH}_3 & \mathsf{O} & \mathsf{OH} \\ \mathsf{H}_3\mathsf{C}-\mathsf{C}-\mathsf{O}-\mathsf{C}-\mathsf{HN}-\mathsf{CH}\cdot\mathsf{CH}_2 \\ \mathsf{CH}_3 & \mathsf{H}_3\mathsf{C}-\mathsf{C}-\mathsf{OH} \\ \mathsf{CH}_3 \\ \mathsf{(I)} \end{array}$$

The molecular structure of (I) is shown in Fig. 1. Compound (I) has a *cis* conformation at the urethane linkage, -O-CO-NH-, and the molecules are linked into a dimer by two $N-H\cdots O=C$ hydrogen bonds (Fig. 2). The *cis* conformation is not common, but it is sometimes observed in short peptides (Oku *et al.*, 2003, 2004; Benedetti *et al.*, 1980). As found in this and other reported structures, the *cis* conformation is suitable for dimerization association in short peptide crystal structures.

Unprotected diols were found as OH groups. From a difference Fourier map, each OH group was found to be disordered around the C–O axis, resulting in two conformers of the OH group. Therefore, there are two possible patterns of hydrogen bonding, each composed of four interactions as shown in Fig. 3. One, shown in green arrows from donors (H atoms) to acceptors (O atoms), has a right-handed orientation: $O22 - H222 \cdots O21^{iii} - H211^{iii} \cdots O21^{v} - H212^{v} \cdots O22^{ii} - H221^{ii} \cdots O22$ (symmetry codes are as in Fig. 3). The other, shown in blue arrows, has a left-handed orientation: $O22 - H222 \cdots O22^{ii} - H221^{ii} \cdots O21^{v} - H211^{v} \cdots O21^{iii} - H212^{iii} \cdots$

Received 4 November 2004 Accepted 22 November 2004 Online 27 November 2004

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Figure 1

A view of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 20% probability level.

O22. Two of the four hydrogen bonds are composed of crystallographically equivalent O atoms, such as $O21^{iii} \cdots O21^{v}$ and $O22^{iii} \cdots O22$. In these pairs, there is a twofold axis between the equivalent O atoms, as shown in Fig. 3. The dimers of (I) connected by two $N-H\cdots O=C$ hydrogen bonds are thus aggregated together by mutual $-OH\cdots OH-$ hydrogen bonds.

Experimental

The title peptide, (I), was prepared from Boc-D-Ser-OH in three steps (overall yield 71%) according to the reported procedure (Dettwiler *et al.*, 2003). Crystals of the title compound were successfully grown from ethyl acetate–hexane. Analytical data (melting point, ¹H NMR and $[\alpha]_D^{20}$) are in accordance with the expected structure; m.p. = 337–338 K, $[\alpha]_D^{20} = +5.1^{\circ}$ (*c* 1.0, methanol).

Crystal data

| 2 | |
|---|---|
| $C_{10}H_{21}NO_4$ $M_r = 219.28$ Orthorhombic, $P2_12_12$ a = 9.833 (8) Å b = 18.994 (17) Å c = 6.713 (8) Å V = 1254 (2) Å ³ Z = 4 | Cu K α radiation Cell parameters from 8702 reflections $\theta = 4.5-66.6^{\circ}$ $\mu = 0.74 \text{ mm}^{-1}$ T = 173.1 K Prism, colorless $0.25 \times 0.25 \times 0.05 \text{ mm}$ |
| $D = 1.162 \text{ Mg m}^{-3}$ | 0.25 × 0.25 × 0.05 mm |
| $D_x = 1.102$ Mg m | |
| Data collection | |
| Rigaku R-AXIS RAPID | 1318 independent reflections |
| diffractometer | 965 reflections with $F^2 > 2\sigma(F^2)$ |
| ω scans | $R_{\rm int} = 0.045$ |
| Absorption correction: refined | $\theta_{\rm max} = 68.1^\circ$ |
| from ΔF (<i>DIFABS</i> ; Walker & | $h = -11 \rightarrow 11$ |
| Stuart, 1983) | $k = -22 \rightarrow 22$ |
| $T_{\min} = 0.765, \ T_{\max} = 0.964$ | $l = -8 \rightarrow 8$ |
| 10 309 measured reflections | |



Figure 2

A packing diagram of (I), projected down the b axis. Hydrogen bonds are shown as dashed lines.



Figure 3

A detail from the packing diagram of (I), projected down the *b* axis. Two hydrogen-bonded states are found in $-OH \cdots OH$ - interactions. One is shown in green arrows (right-handed orientation); the other is shown in blue arrows (left handed). [Symmetry codes: (ii) -x, 1 - y, z; (iii) x, y, 1 + z; (v) -x, 1 - y, 1 + z.]

Refinement

| Refinement on F^2 | H-atom parameters constrained |
|---------------------------------|--|
| $R[F^2 > 2\sigma(F^2)] = 0.085$ | $w = 1/[0.007F_o^2 + \sigma(F_o^2)]/(4F_o^2)$ |
| $wR(F^2) = 0.244$ | $(\Delta/\sigma)_{\rm max} < 0.001$ |
| S = 1.07 | $\Delta \rho_{\rm max} = 0.59 \ {\rm e} \ {\rm \AA}^{-3}$ |
| 1318 reflections | $\Delta \rho_{\rm min} = -0.43 \text{ e } \text{\AA}^{-3}$ |
| 133 parameters | |

Table 1

Torsion angles for the urethane linkage (°).

| C11-O11-C15-N21 | 175.5 (5) | C15-N21-C21-C22 | -135.8 (6) |
|-----------------|-----------|-----------------|------------|
| C21-N21-C15-O11 | 4.8 (8) | C15-N21-C21-C23 | 97.8 (6) |

| Table 2 | |
|-----------------------------------|--|
| Hydrogen-bonding geometry (Å, °). | |

| $D - H \cdot \cdot \cdot A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - \mathbf{H} \cdots A$ |
|------------------------------|------|-------------------------|--------------|---------------------------|
| $N21-H1\cdots O12^{i}$ | 0.95 | 1.97 | 2.916 (6) | 174 |
| $O21 - H211 \cdots O21^{ii}$ | 0.78 | 2.15 | 2.717 (8) | 129 |
| $O22-H222\cdots O21^{iii}$ | 0.82 | 2.11 | 2.742 (6) | 133 |
| $O21 - H212 \cdots O22^{iv}$ | 0.82 | 2.11 | 2.742 (6) | 134 |
| $O22-H221\cdots O22^{ii}$ | 0.85 | 1.95 | 2.799 (6) | 177 |

Symmetry codes: (i) 1 - x, 1 - y, z; (ii) -x, 1 - y, z; (iii) x, y, 1 + z; (iv) x, y, z - 1.

The methyl C atoms were refined isotropically; all other non-H atoms were refined anisotropically. This procedure was adopted because of the limited number of observed reflections. H atoms were positioned geometrically, with C-H = 0.95 Å, except for hydroxy H atoms H211, H212, H221 and H222, which were located in a difference Fourier map with occupancies of 0.5. The H atoms were refined using a riding model, with U_{iso} constrained to be $1.2U_{eq}$ of the carrier atom. In the absence of significant anomalous scattering effects, Friedel pairs were averaged and the absolute configuration could not be determined from the diffraction experiment. The absolute configuration of the compound was, however, confirmed from the spectroscopic data.

Data collection: *RAPID-AUTO* (Rigaku/MSC & Rigaku Corporation, 2003); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC & Rigaku Corporation, 2003); program(s) used to solve structure: *SIR2002* (Burla *et al.*, 2003); program(s) used to refine structure: *CRYSTALS* (Watkin *et al.*, 1996); molecular graphics: *ORTEP* (Johnson, 1965); software used to prepare material for publication: *CrystalStructure*. HO acknowledges a Grant-in-Aid for Scientific Research on Priority Areas (No. 14078101 and 16033211, Reaction Control of Dynamic Complexes) from the Ministry of Education, Culture, Sports, Science and Technology, Japan.

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